

Molecular Interface Zone (phase?) Structures and Contact Angles on Organic Solids

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I. Molecular Structure Gradients at Liquid-Solid Interfaces:

- “Acid-Base” + Dispersion. Functionally-terminated C-16 SAMs + pendant-group polymers.
- Inverse adhesion work: donor-acceptor vs dispersion correlations. electronic density shifts and amphoteric structure gradients?
- SAMs as models for polymers?

II. Wetting Correlations in Controlled Coverage Alkyl Chain Monolayers:

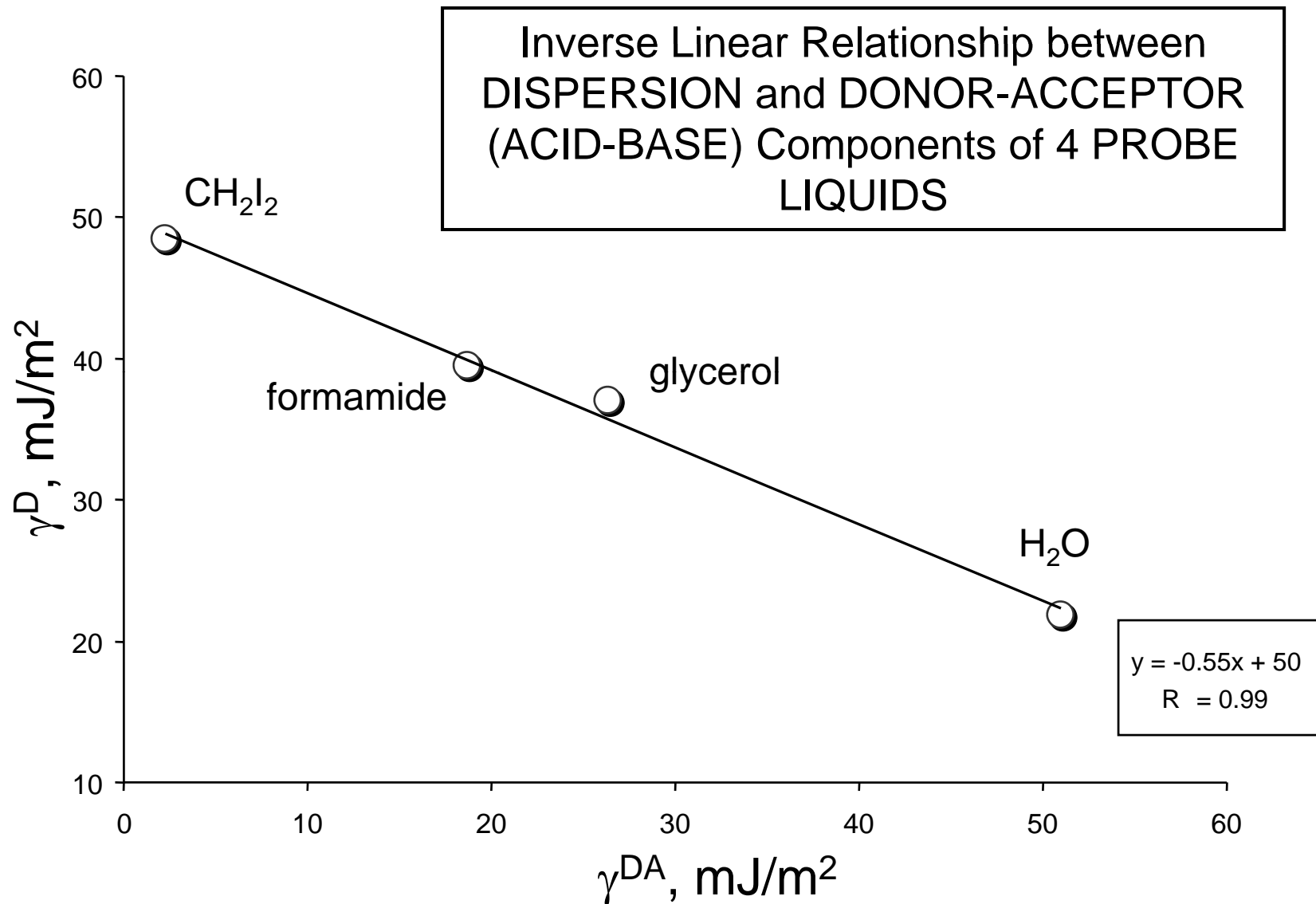
Dispersion Forces Wetting as a measure of surface phases in CH₃-terminated C-18 SAMs?

Major points in wetting (already appreciated by many)

- 1. Wetting (molecular liq/solid) interfaces are not sharp at molecular scale**
- 2. Transition zones appear at interface**
- 3. Relaxation in transition zone for soft structures**
 - Nuclear restructuring**
 - Position**
 - Orientation**
 - Electron density shifts**
- 4. Inverse relationship (conjugate) between long range, polarization (dispersion) and short range (chemical / donor-acceptor) forces**

**Interesting observation of literature
wetting data**

**“inverse”
dispersion – donor acceptor
correlation**

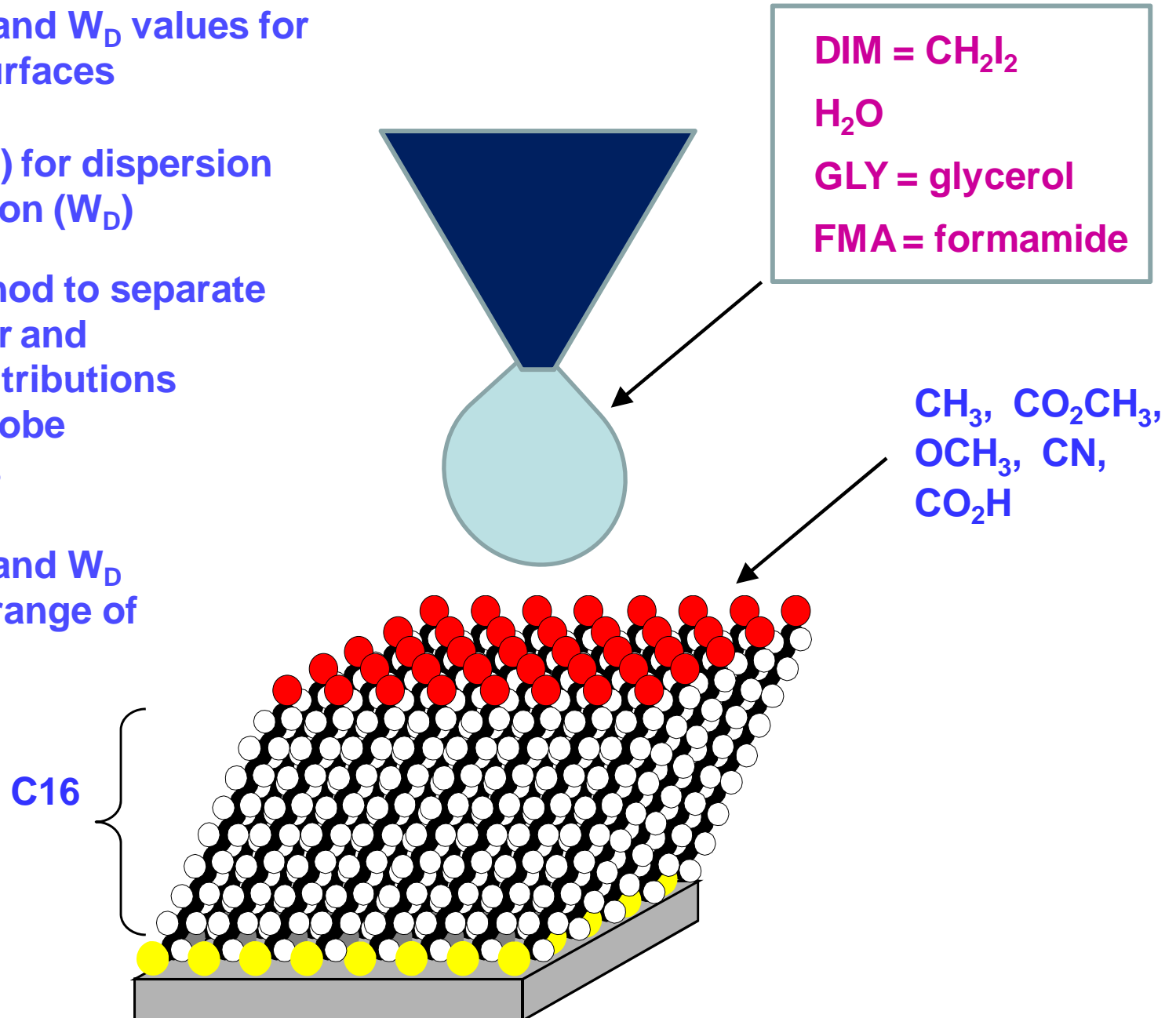


Data from, Fowkes in D.H. Kaelble, Physical Chemistry of Adhesion, Wiley-Interscience, New York, 1971

**Using Well-Defined Terminally Functionalized
Self-Assembled Monolayers (SAMs) to Test the
Inverse Relationship**

Strategy

- Correlate W_{AB} and W_D values for the range of surfaces
- Use CH_2I_2 (DIM) for dispersion work of adhesion (W_D)
- Proposed method to separate donor-acceptor and dispersion contributions using other probe measurements
- Correlate W_{AB} and W_D values for the range of surfaces



Data Analysis for SAMs

SAM End Group (X)	Total Work of Adhesion (W_A), mJ/m ²			DIM	W_A^*	δW_A^* , mJ·m ⁻²
	<u>H₂O</u>	<u>GLY</u>	<u>FMA</u>			
CH ₃				68.3	1.000	0.0
OCH ₃				80.8	1.183	12.5
CO ₂ CH ₃				81.2	1.189	12.9
CN				93.1	1.363	24.8
CO ₂ H				93.1	1.363	24.8

CH₂I₂, pure dispersion liq

$$\equiv \frac{W_A(\text{DIM}/X)}{W_A(\text{DIM}/\text{CH}_3)}$$

$$\delta W_A^*, \text{mJ}\cdot\text{m}^{-2}$$

“extra” dispersion interactions

normalized W_A
 (relative to pure disp. liq on pure disp. solid)

Data Analysis for SAMs

Total Work of Adhesion (W_A), mJ/m²

<u>SAM End Group (X)</u>	<u>H₂O</u>	<u>GLY</u>	<u>FMA</u>	<u>DIM</u>	<u>W_A^*</u>
CH ₃	48.3	61.3	49.6	68.3	1.000
OCH ₃	85.7	76.3	75.4	80.8	1.183
CO ₂ CH ₃	95.7	83.2	83.9	81.2	1.189
CN	103.3	96	93.4	93.1	1.363
CO ₂ H	146.5	119.8	116	93.1	1.363

CH₂I₂, pure dispersion liq

$\equiv \frac{W_A(\text{DIM}/X)}{W_A(\text{DIM}/\text{CH}_3)}$

Dispersion + Donor-Acceptor (DA)

normalized W_A (relative to pure disp. liq on pure disp. solid)

"extra" dispersion

Use W_A^* to Break W_A for Each Liquid into The “Core” Dispersion Component and the Conjugate Donor-Acceptor Component for SAM Terminal Group X

1.) Define $W_A^{CD}(\text{liq}/X)$

$$\frac{W_A^{CD}(\text{liq}/X)}{W_A(\text{liq}/X)} = \frac{W_A(\text{DIM}/\text{CH}_3)}{W_A(\text{DIM}/X)} \equiv W_A^*$$

$$W_A^{CD}(\text{liq}/X) \equiv \frac{W_A(\text{liq}/X)}{W_A^*}$$

2.) Assert: $W_A^{CD}(\text{liq}/X) \sim W_A^D(\text{liq}/X)$

3.) Calculate $W_A^{DA}(\text{liq}/X)$

$$W_A(X) = W_A^{CD}(\text{liq}/X) + W_A^{DA}(\text{liq}/X)$$

↑
Conjugate DA component

Results

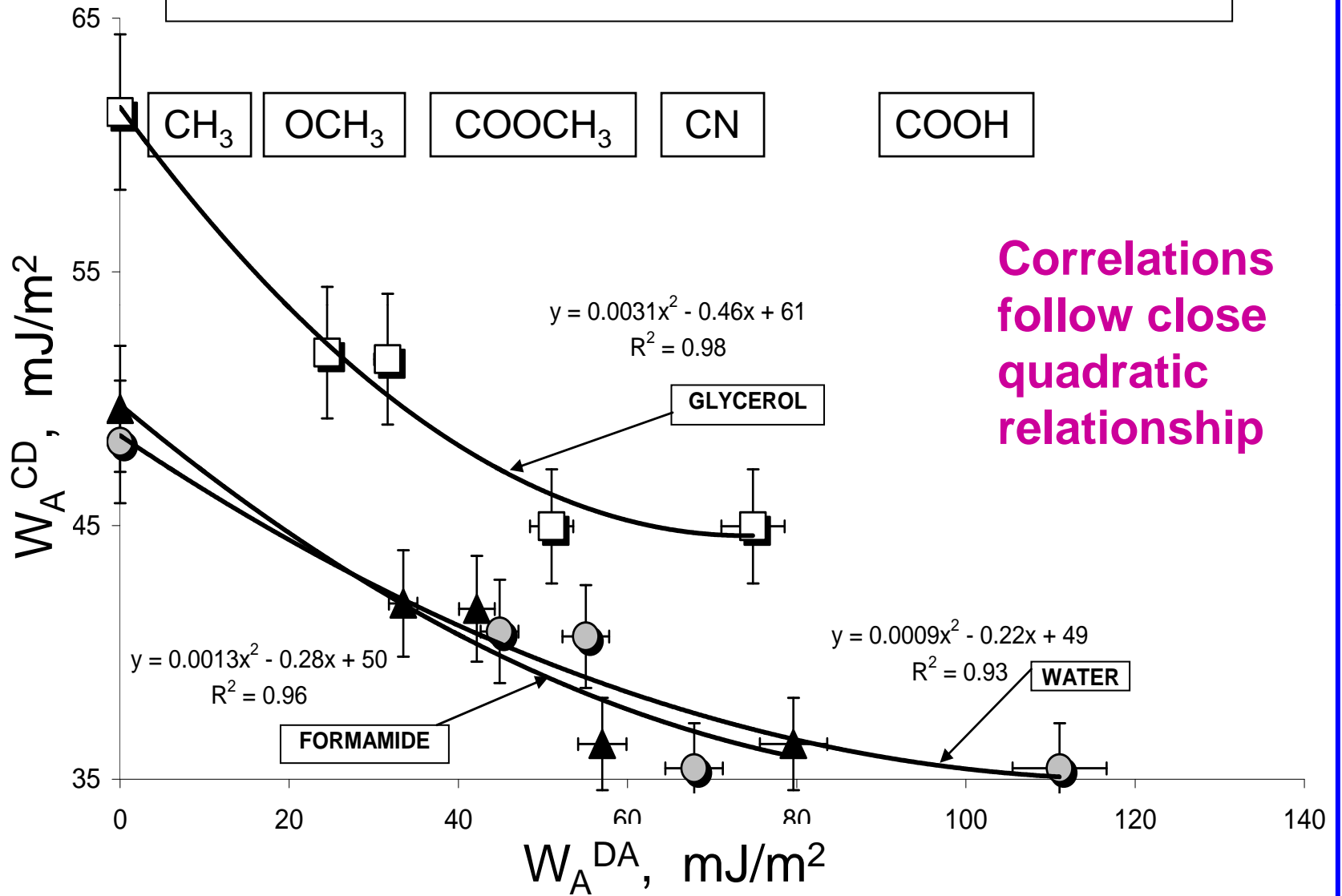
“Core” Dispersion Component
 $[W_A^{CD} (\text{liq}/X)]$ (mJ/m²)

<u>X</u>	<u>H₂O</u>	<u>GLY</u>	<u>FMA</u>
CH ₃	48.3	61.3	49.6
OCH ₃	40.8	51.8	41.9
CO ₂ CH ₃	40.6	51.6	41.7
CN	35.4	45.0	36.4
CO ₂ H	35.4	45.0	36.4

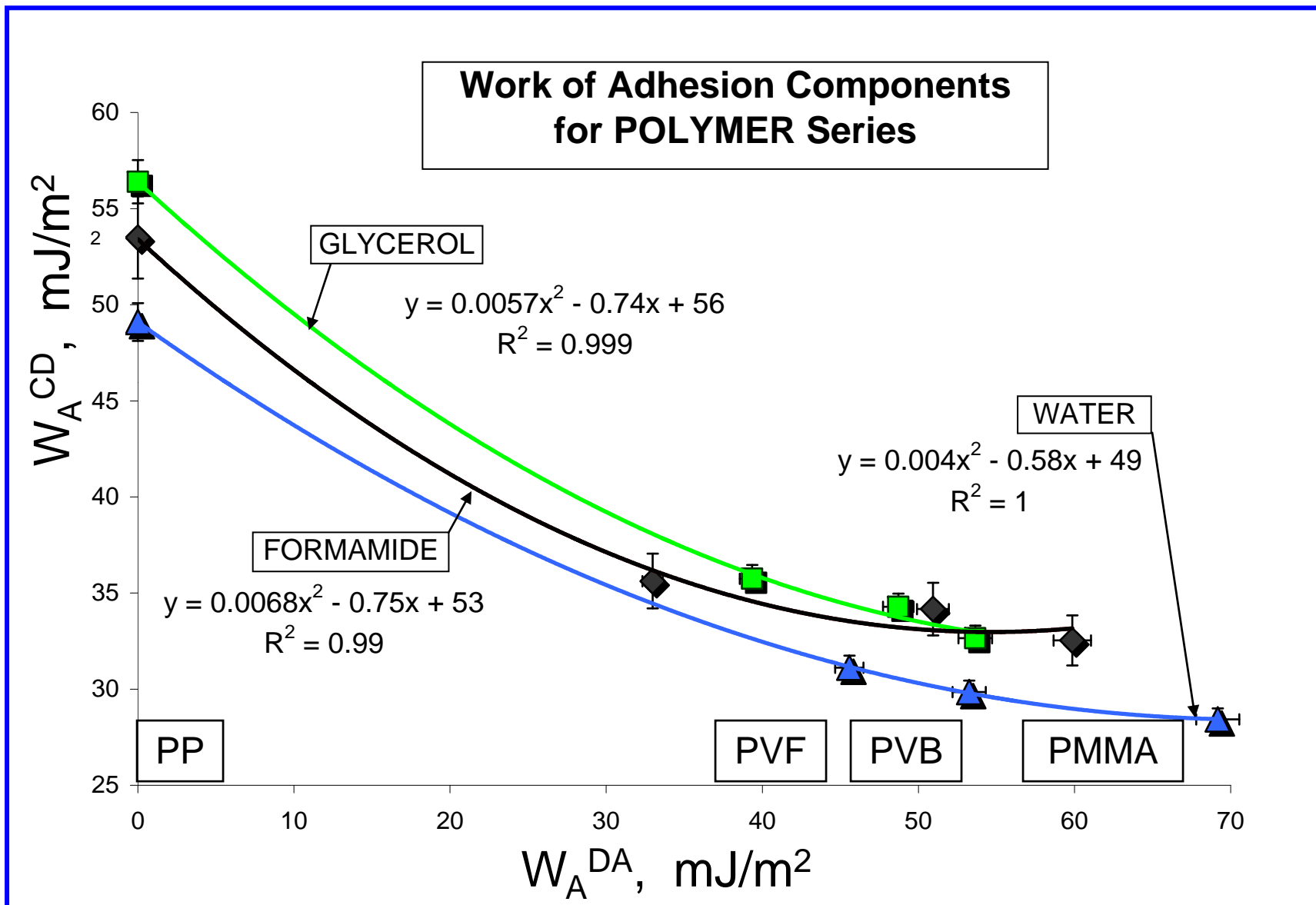
Component due to X capable of
Donor-Acceptor Components
 $[W_A^{DA} (\text{liq}/X)]$ (mJ/m²)

<u>H₂O</u>	<u>GLY</u>	<u>FMA</u>
0.0	0.0	0.0
44.9	24.5	33.5
55.1	31.6	42.2
67.9	51.0	57.0
111.1	74.8	79.6

DISPERSION *versus* DONOR-ACCEPTOR COMPONENTS
of **SAM** WORK of ADHESION

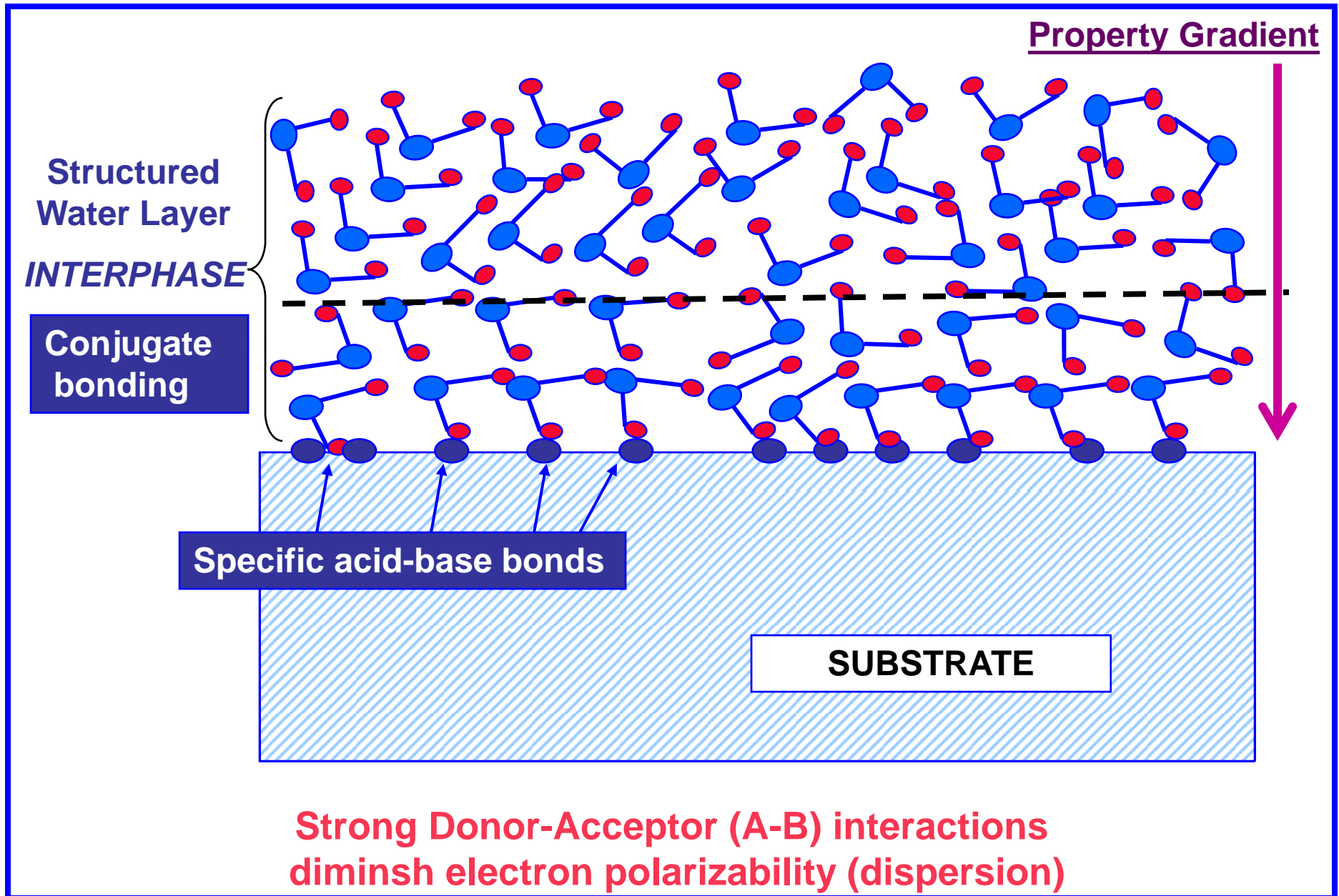


Using literature values for polymers



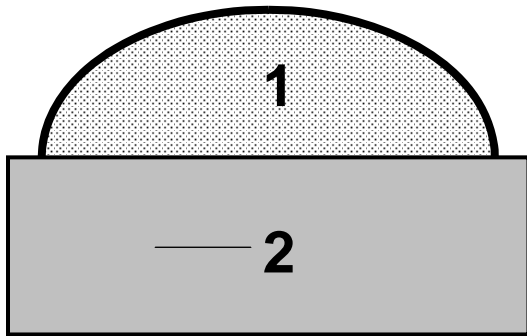
Data taken from: C. Della Volpe, et al., *J. Adhesion Sci. Technol.* **17**, 1425 (2003), and reanalyzed.

Interface Zone Nuclear and Electronic Gradients



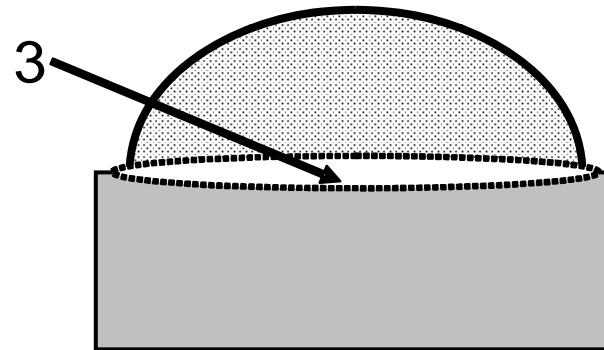
HAMAKER CONSTANTS

Interface line



$$A_{12} = (A_{11} \times A_{22})^{1/2}$$

Molecular Interface Zone



$$A_{123} = (A_{11}^{1/2} - A_{33}^{1/2})(A_{22}^{1/2} - A_{33}^{1/2})$$

CONCLUSIONS

- Interfacial ZONE exists in L/S & L/L cases
- unique structure, electron density and nuclear properties from spectroscopy (literature --- SFG, SFG, etc.)
- Often an average between bulk phases
- Hamaker relations w/A_{33} are 1/quadratic
- W^D shows 1/quadratic relationship to W^{DA} for both SAMs and Polymers